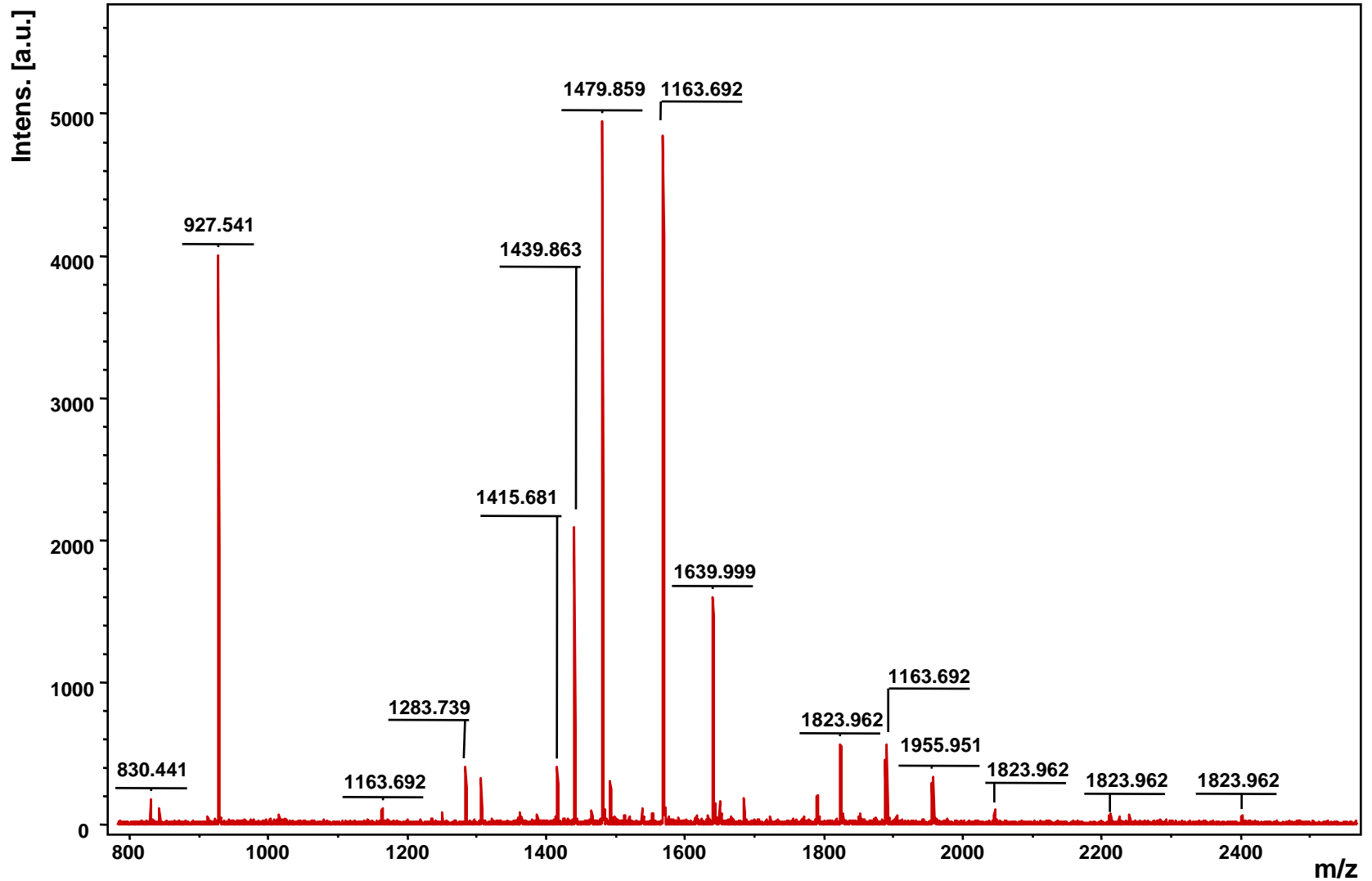


ESPECTROMETRIA DE MASAS

CARACTERISTICAS DE LA ESPECTROMETRIA DE MASAS COMO TECNICA ANALITICA

- POSIBILITA LA DETERMINACION DE LA RELACION MASA/CARGA (M/Z) DE UN ION EN UNA MEZCLA Y EN CONSECUENCIA SU MASA MOLECULAR, ej. El pm de un péptido o una proteína.**
- OFRECE INFORMACION ESTRUCTURAL DE UN ION SELECCIONANDO ANALIZANDO LAS MASAS DE LOS FRAGMENTOS QUE SE PRODUCEN EN LA COLISION ENTRE ESTE ION Y LAS MOLECULAS DE UN GAS NEUTRO, ej. La secuencia de aminoácidos de un péptido o modificaciones en su estructura primaria.**

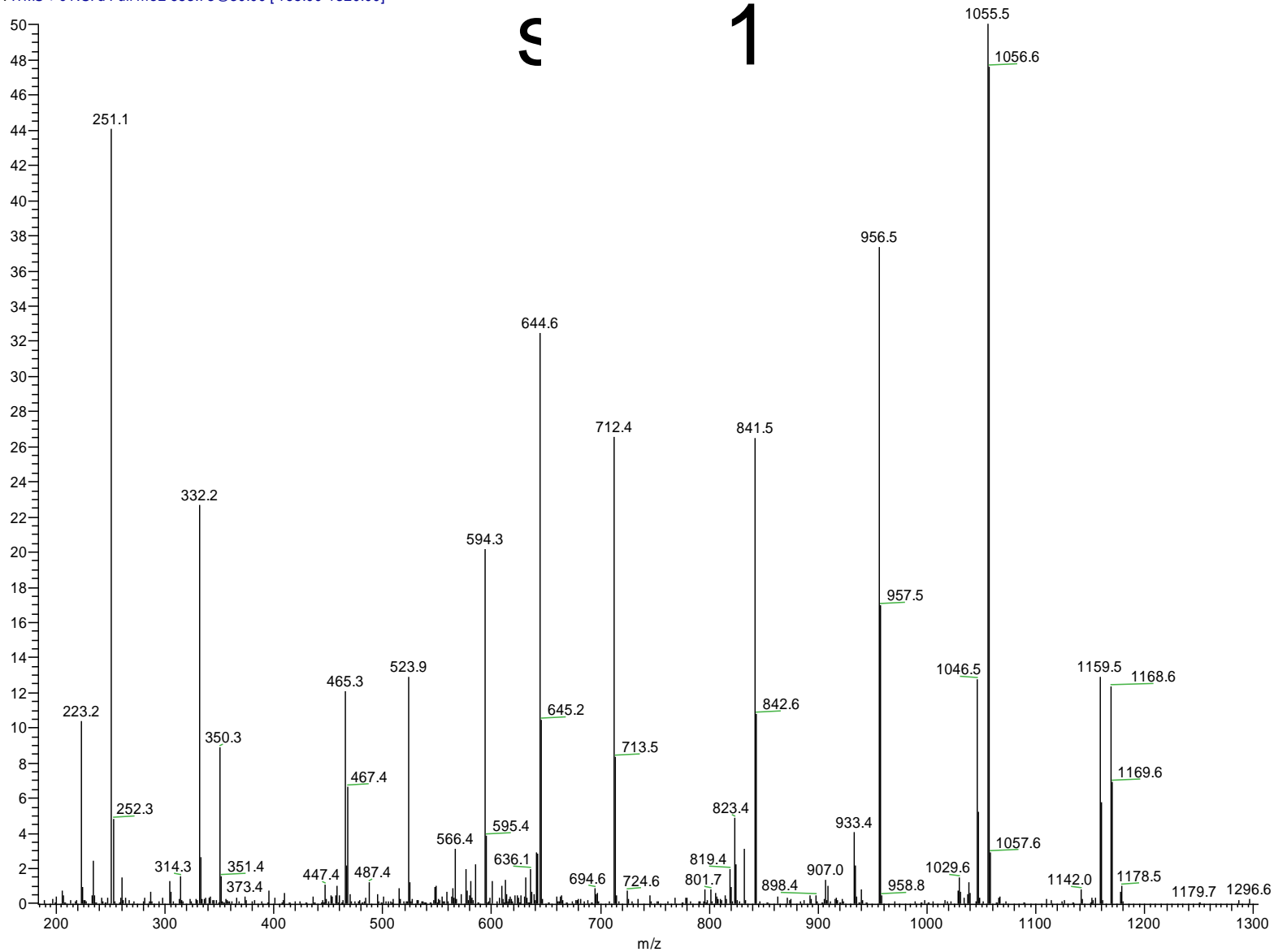
Mapa MALDI-TOF de un digerido en gel de BSA



HLVDEPQNLK

BSA_031203_NEWDIGEST02 #4327-4332 RT: 18.43-18.44 AV: 2 NL: 3.49E3
T: ITMS + c NSI d Full ms2 653.73@30.00 [165.00-1320.00]

ε 1



m/z Value

$$m/z = (M + nH)/n \approx (M + n)/n$$

$$M = (m/z) n - nH \approx (m/z) n - n$$

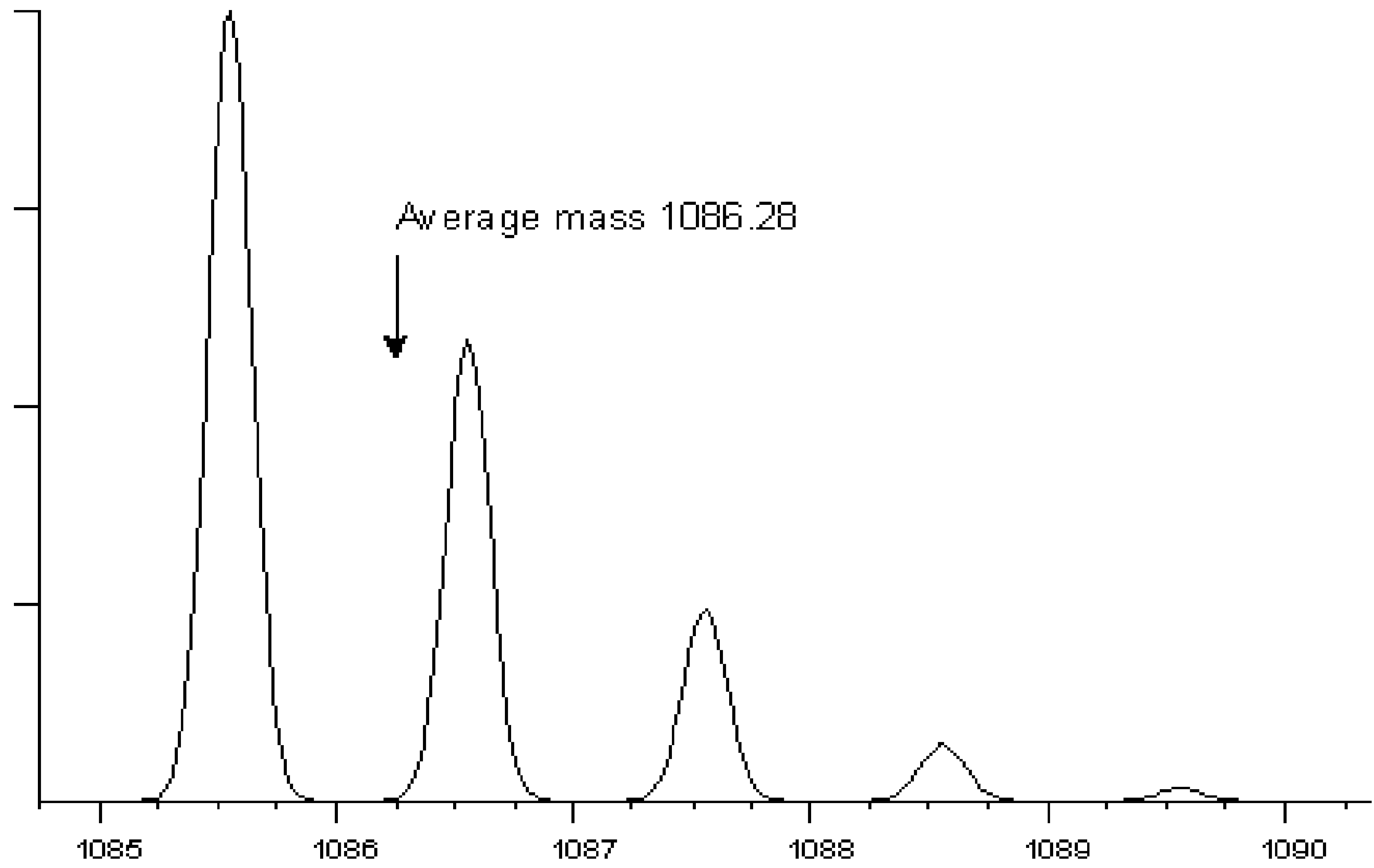
M: mass of the measured molecule

n: number of H⁺

H: H⁺ mass

Monoisotopic mass 1085.55

Average mass 1086.28

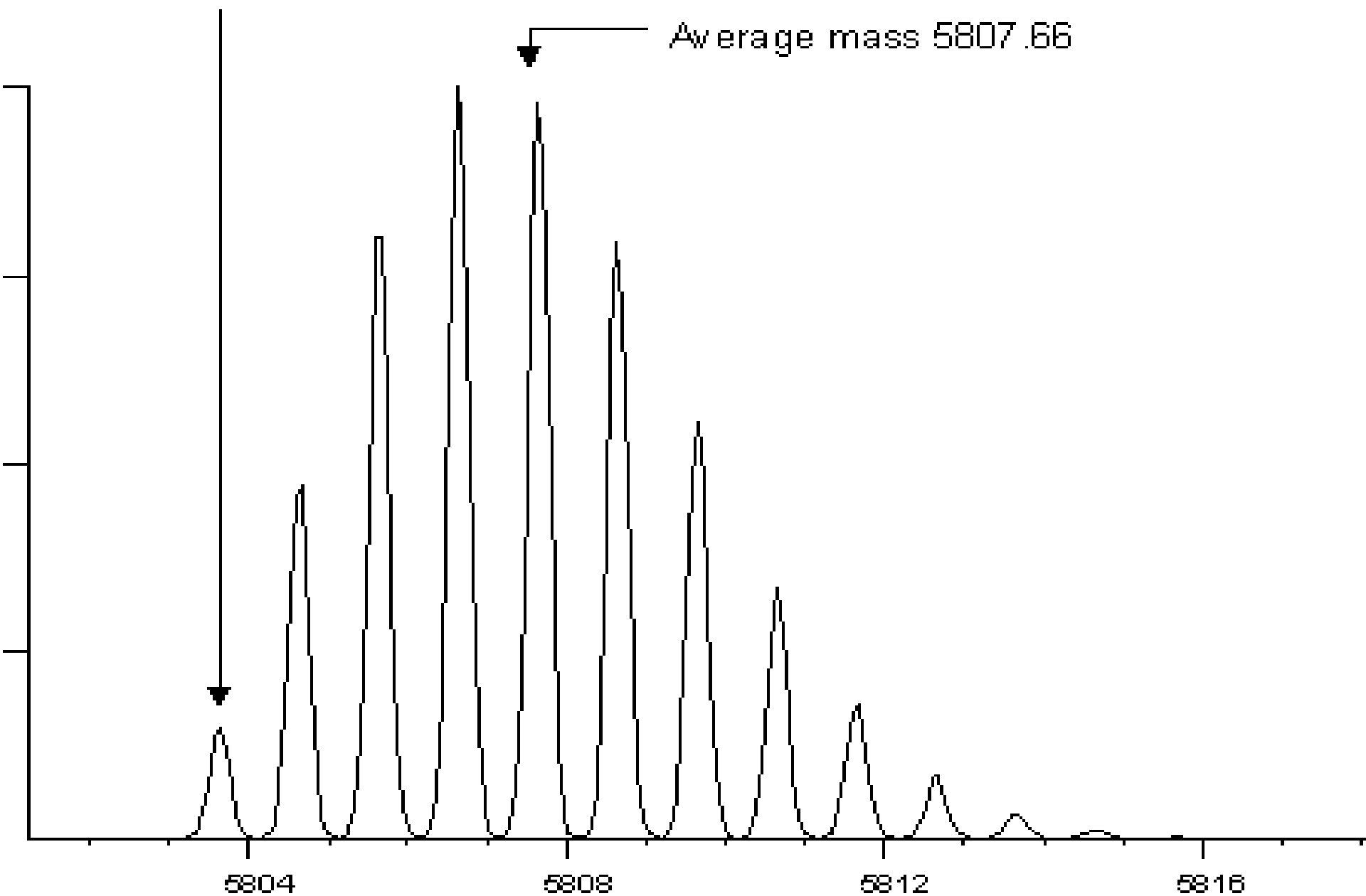


Isotopic abundances of some common elements

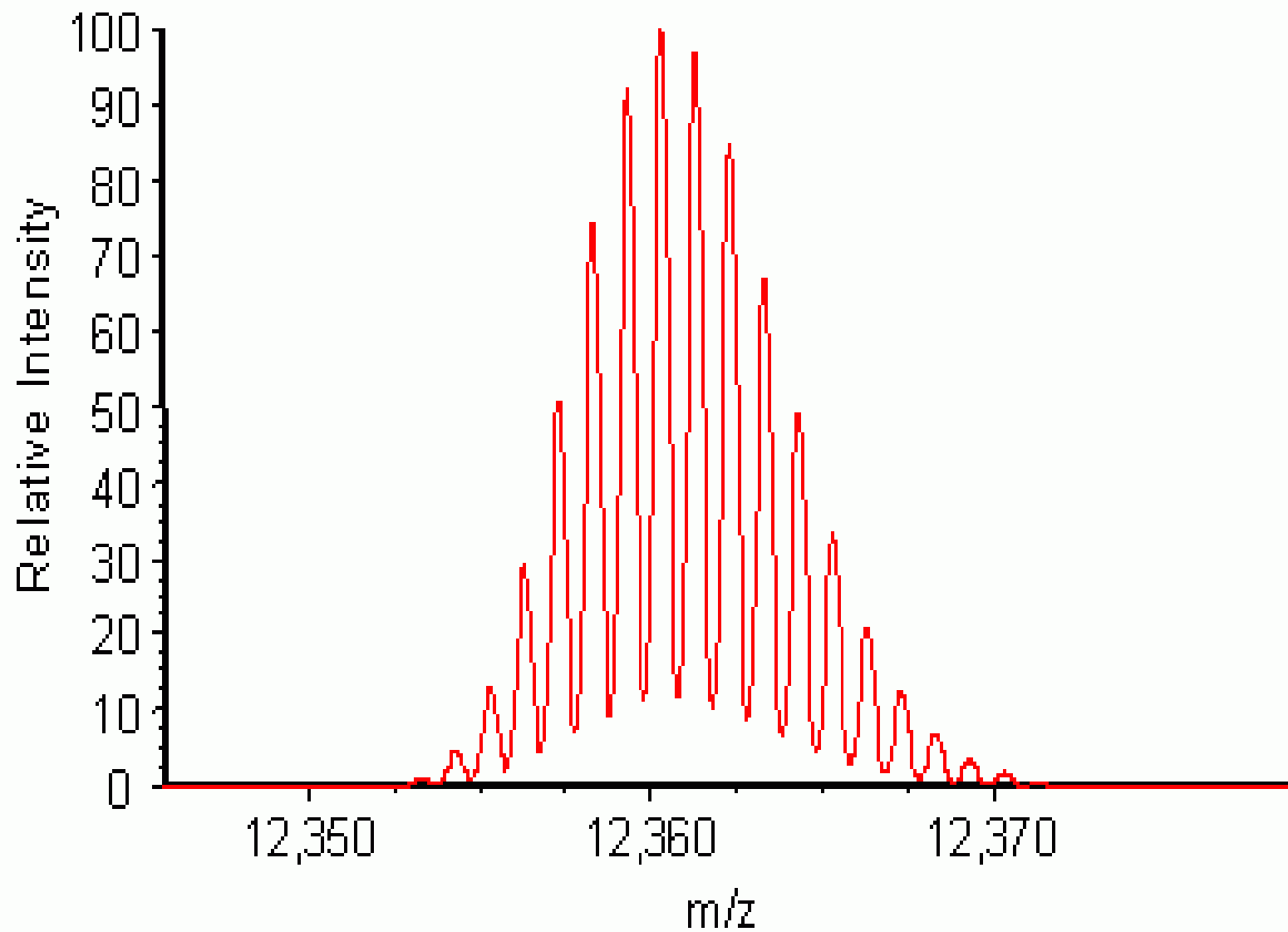
Element	Mass	Natural Abundance
H	1.0078	99.985%
	2.0141	0.015
C	12.0000	98.89
	13.0034	1.11
N	14.0031	99.64
	15.0001	0.36
O	15.9949	99.76
	16.9991	0.04
	17.9992	0.20
F	18.9984	100
S	31.9721	95.00
	32.9715	0.76
	33.9679	4.22
	35.9671	0.02

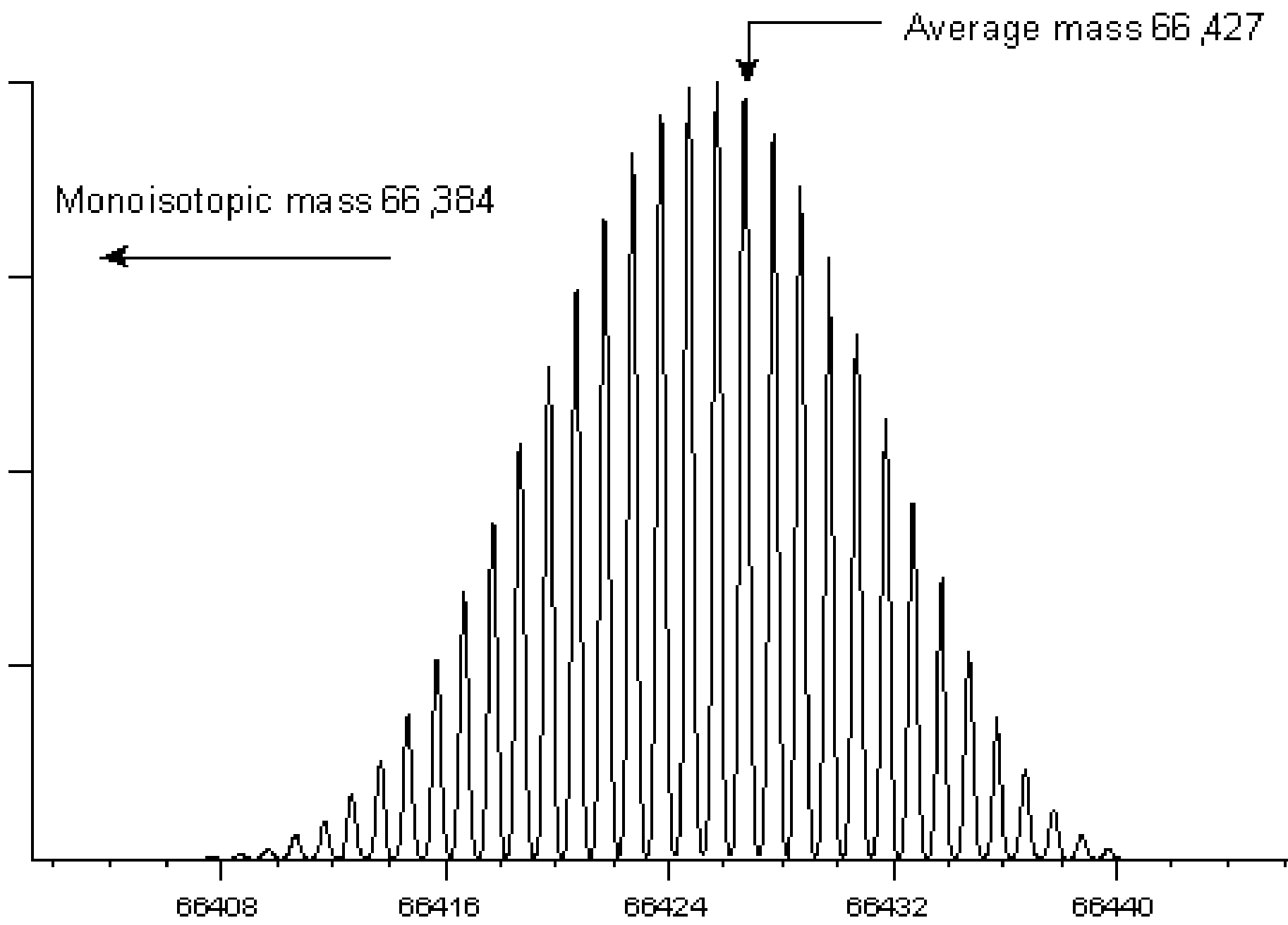
Monoisotopic mass 5803.64

Average mass 5807.66

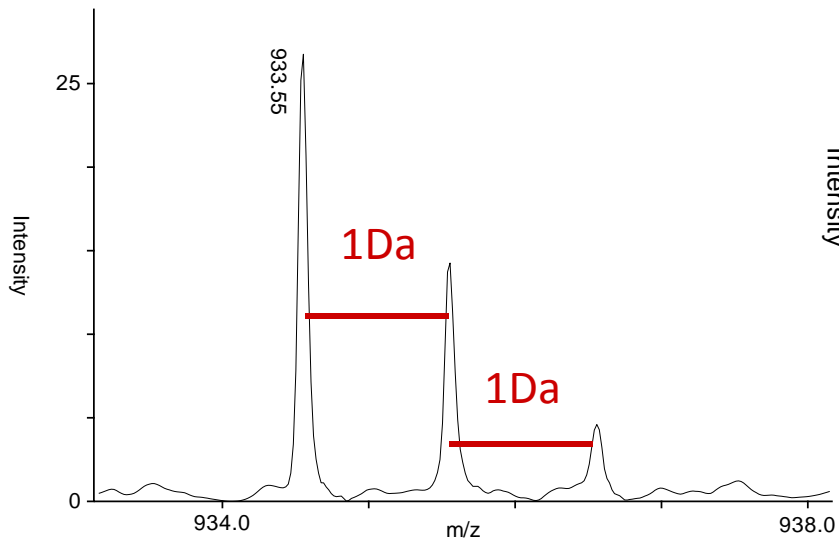


Cytochrome C (horse) at 25,000 m/Δm

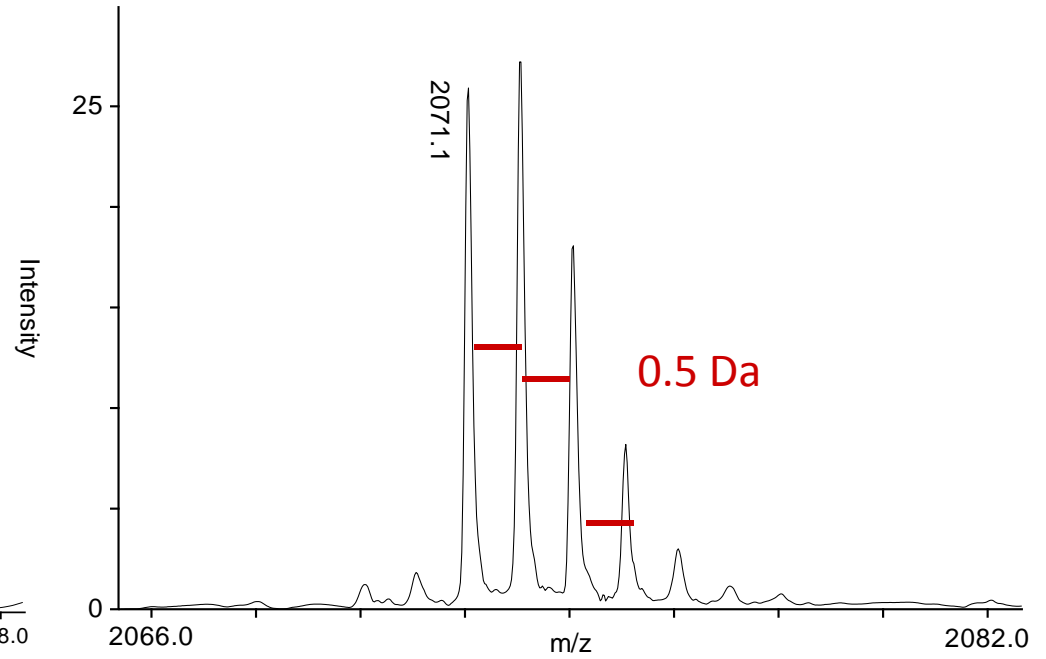




Isotopic cluster



C:\Documents and Settings\VMCosta\Ambiente de trabalho\alex\AlexGva10.txt (20:00 12/29/04)



C:\Documents and Settings\VMCosta\Ambiente de trabalho\alex\AlexGva10.txt (20:02 12/29/04)

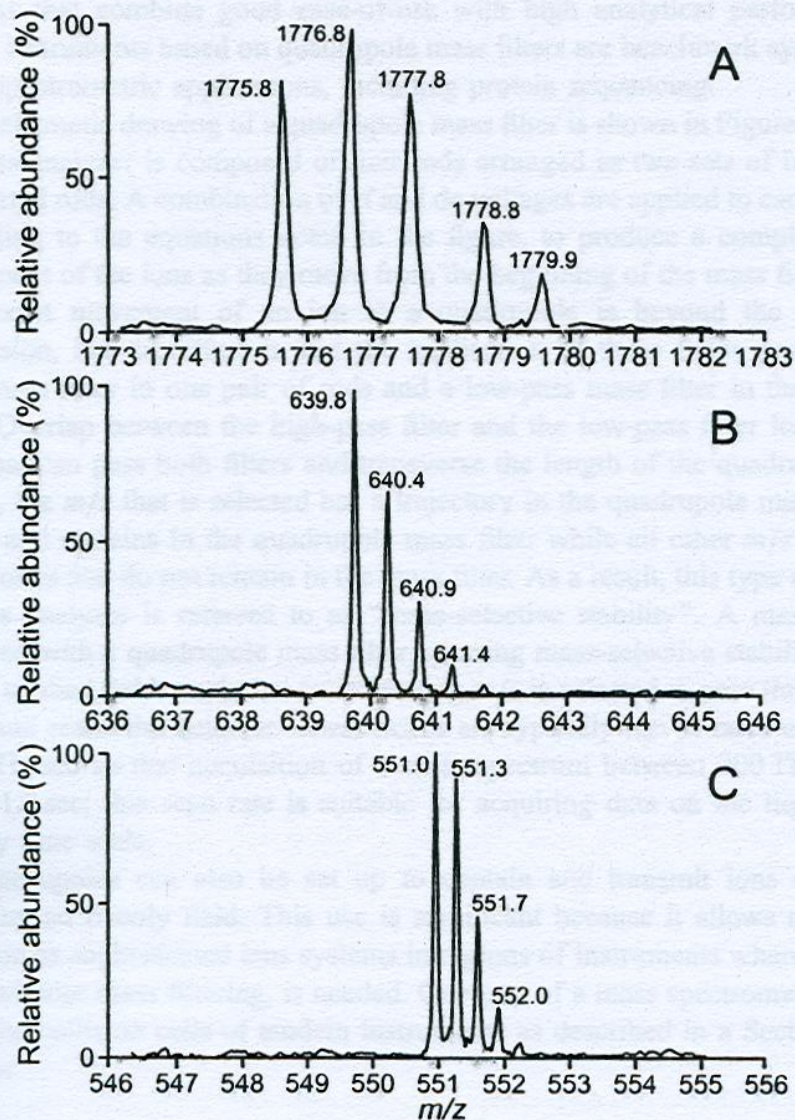
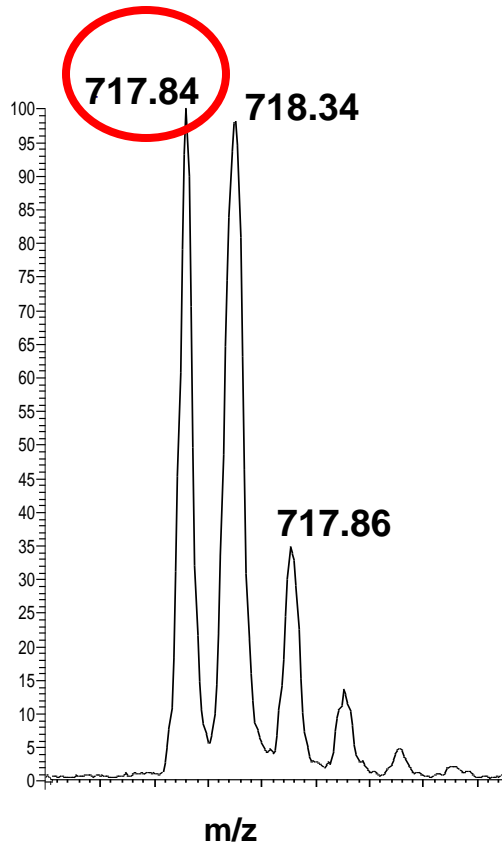
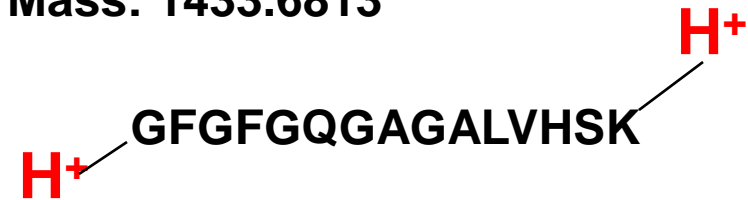


figure 3.7. Resolved isotope clusters of peptide ions at different charge states. Resolution of the isotope cluster of a peptide ion allows the charge state of that ion to be determined. (A) The isotope cluster for a singly charged peptide ion ($M + H$)⁺ is characterized by an m/z difference of 1.0 Th. (B) The isotope cluster for a doubly charged peptide ($M + 2H$)²⁺ is characterized by an m/z difference of 0.5 Th. (C) The isotope cluster of a triply charged peptide ($M + 3H$)³⁺ is characterized by an m/z difference of 0.33 Th.

m/z Value



Elemental Composition: C₆₄ H₉₃ N₁₈ O₂₀
Monoisotopic Mass: 1433.6813



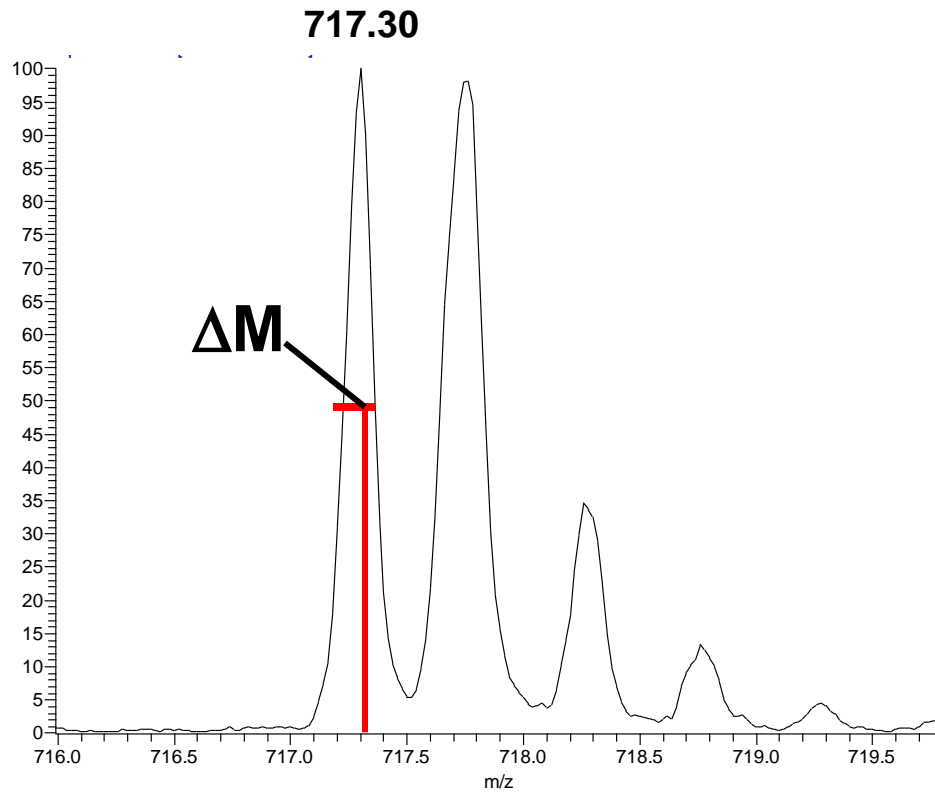
$$m/z = (M + nH)/n \approx (M + n)/n$$

$$m/z = (1433.6813 + 2H)/2$$

$$\approx (1433.6813 + 2)/2$$

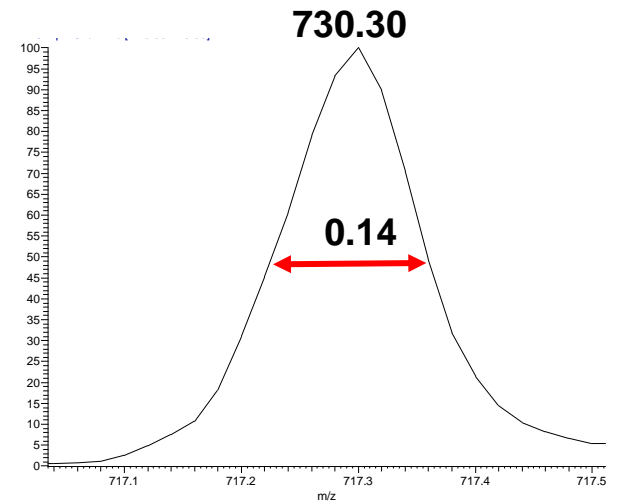
$$m/z = 717.8406$$

Resolution



Resolution: $M/\Delta M$

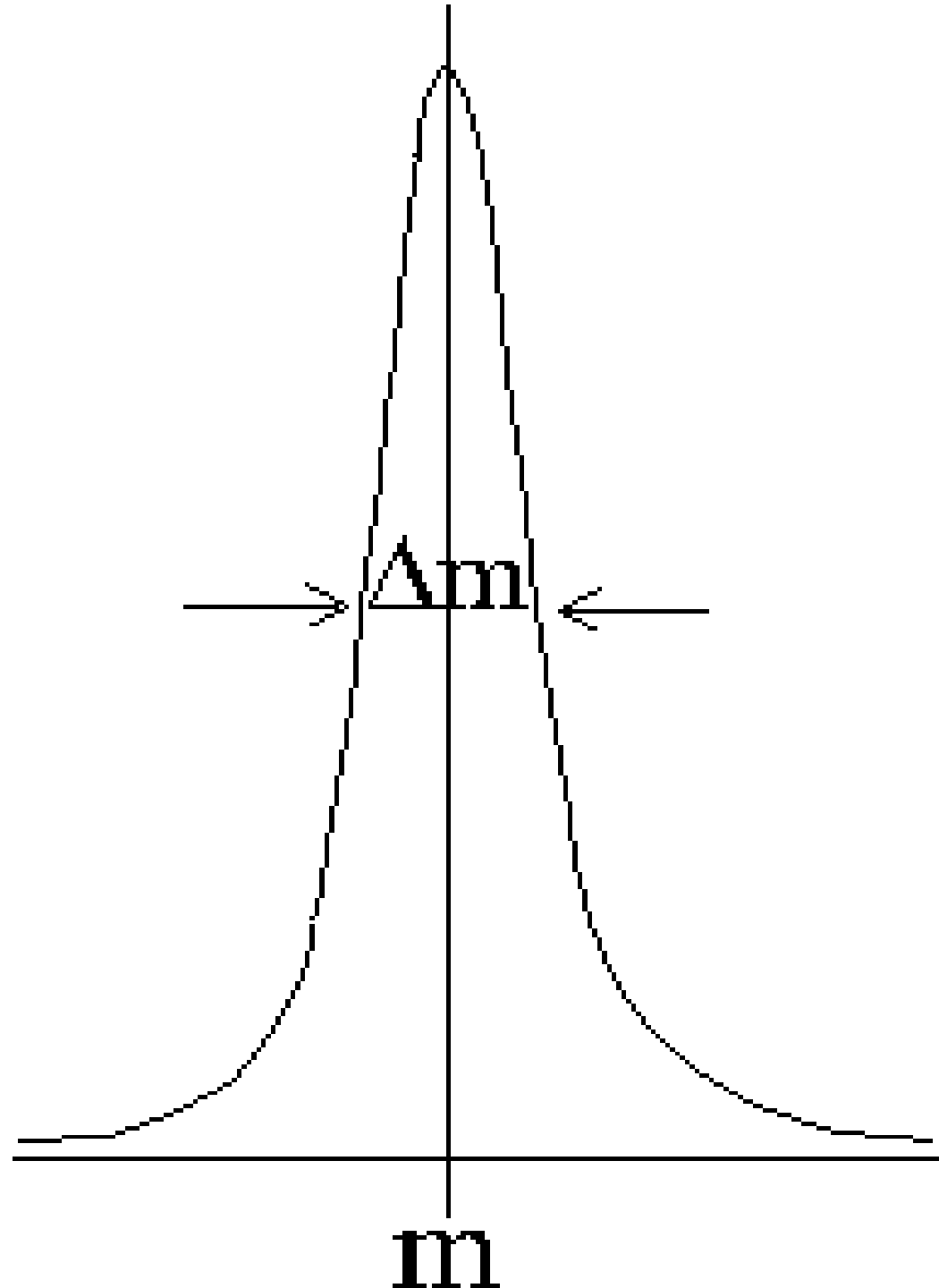
$$\text{Resolution} = 730.30/0.14$$
$$= 5216.42$$



RESOLUCION:

$$R = M/\Delta M$$

Cuando se mide
la anchura de
pico a altura
media: FWHM



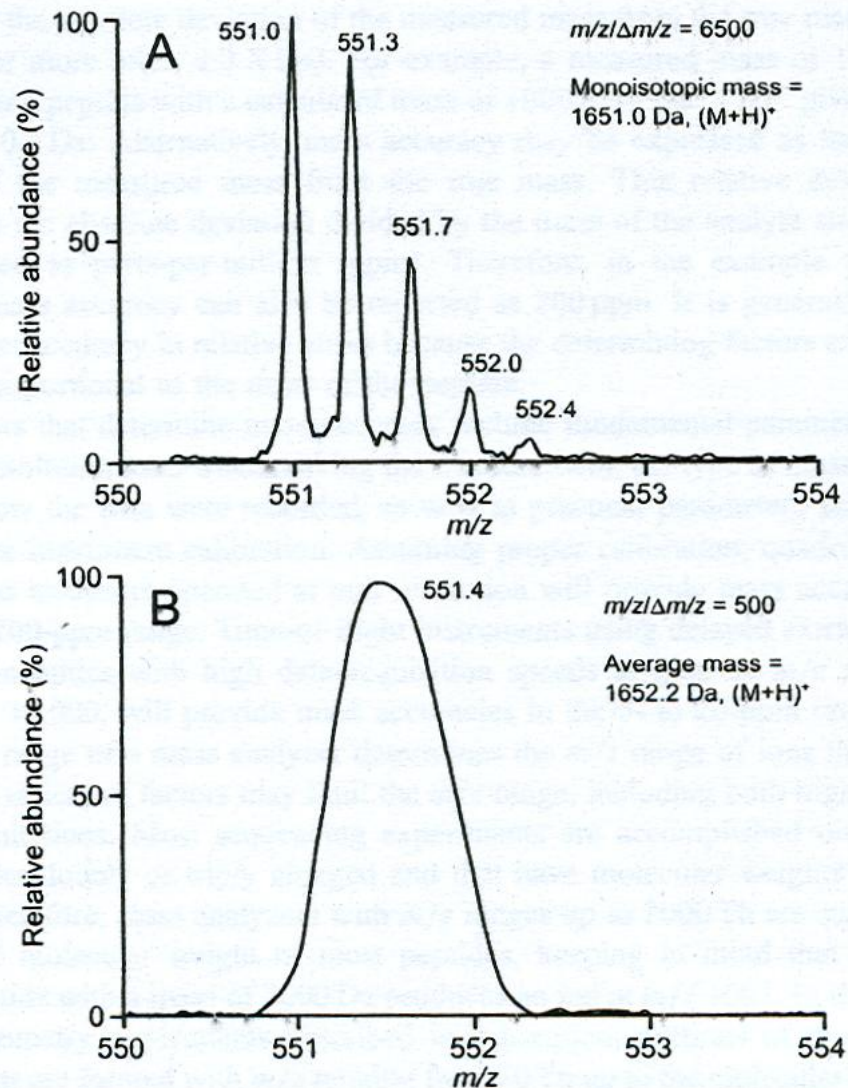
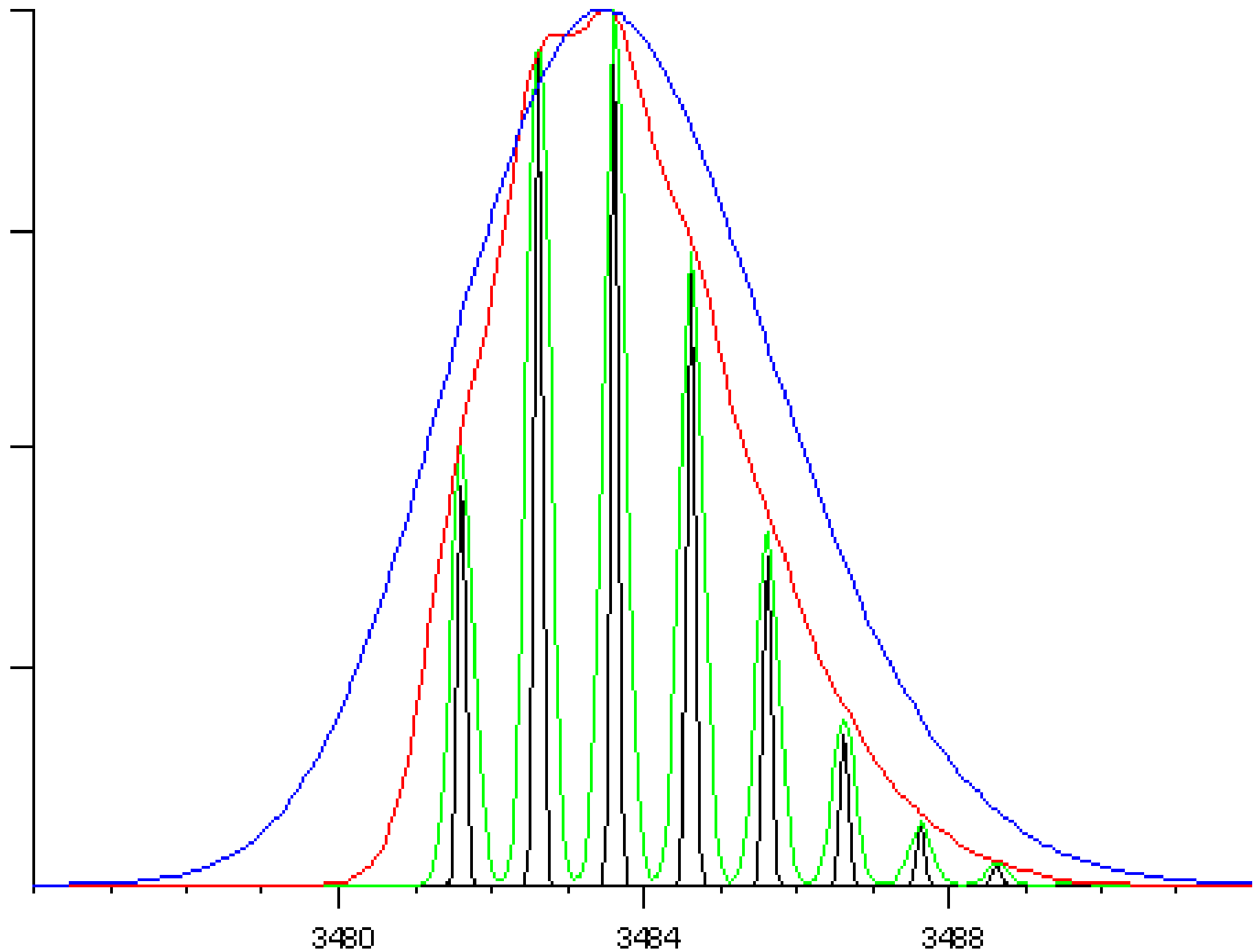
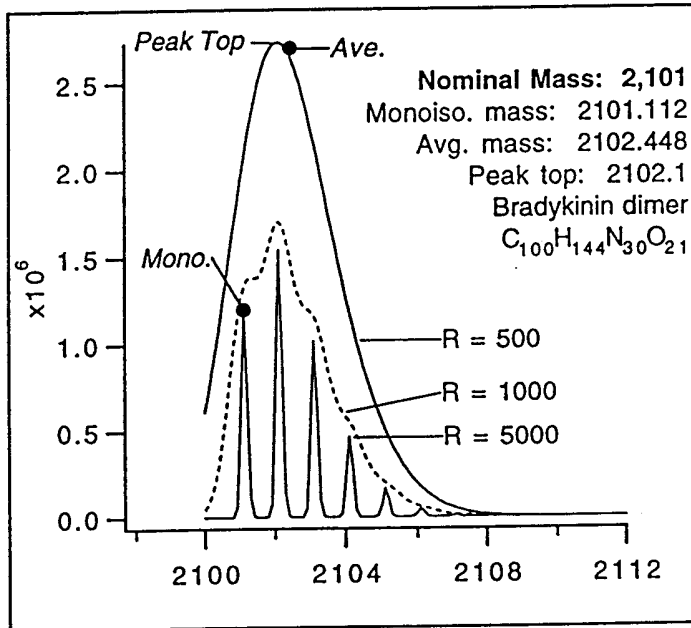
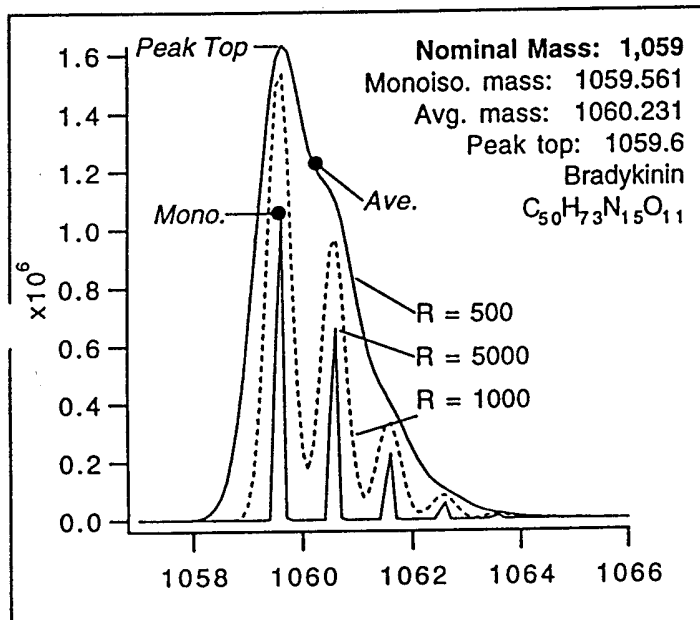
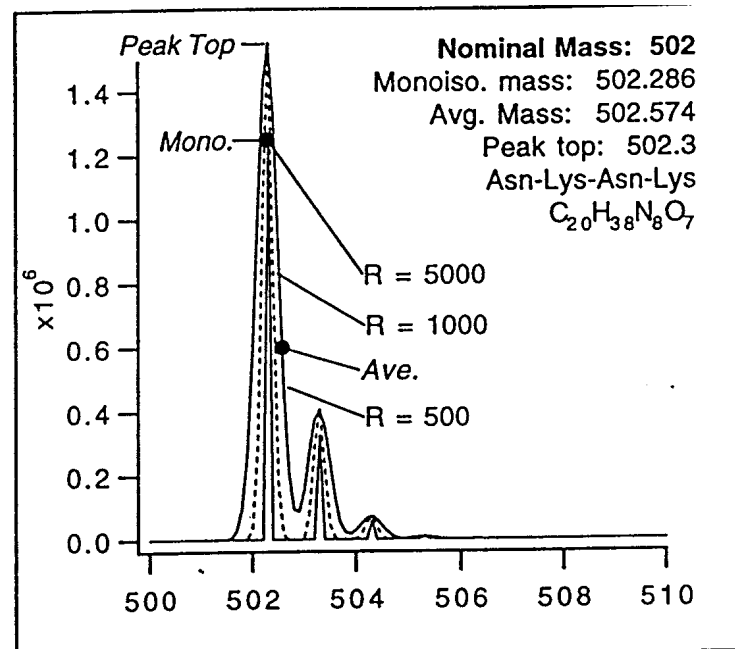
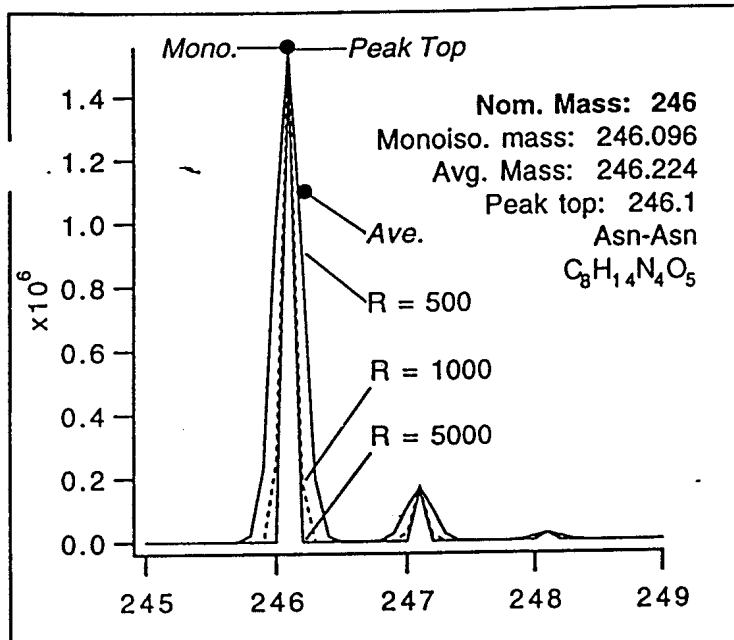
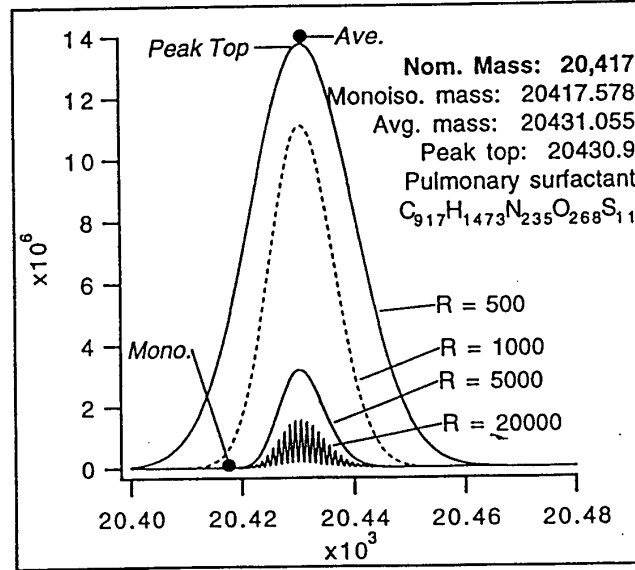
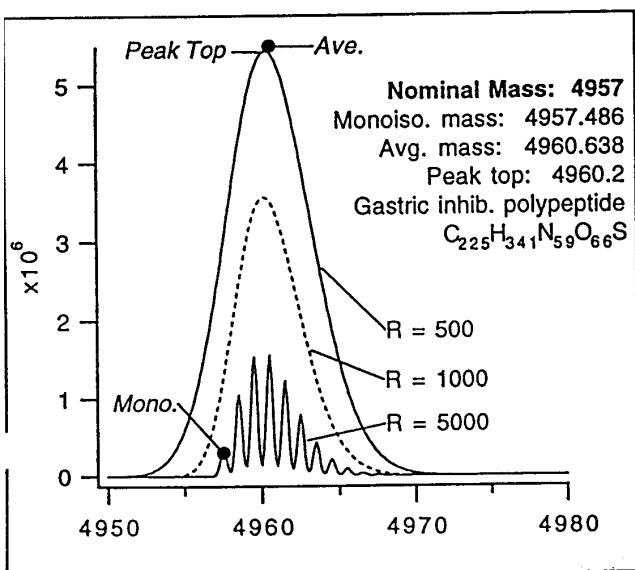


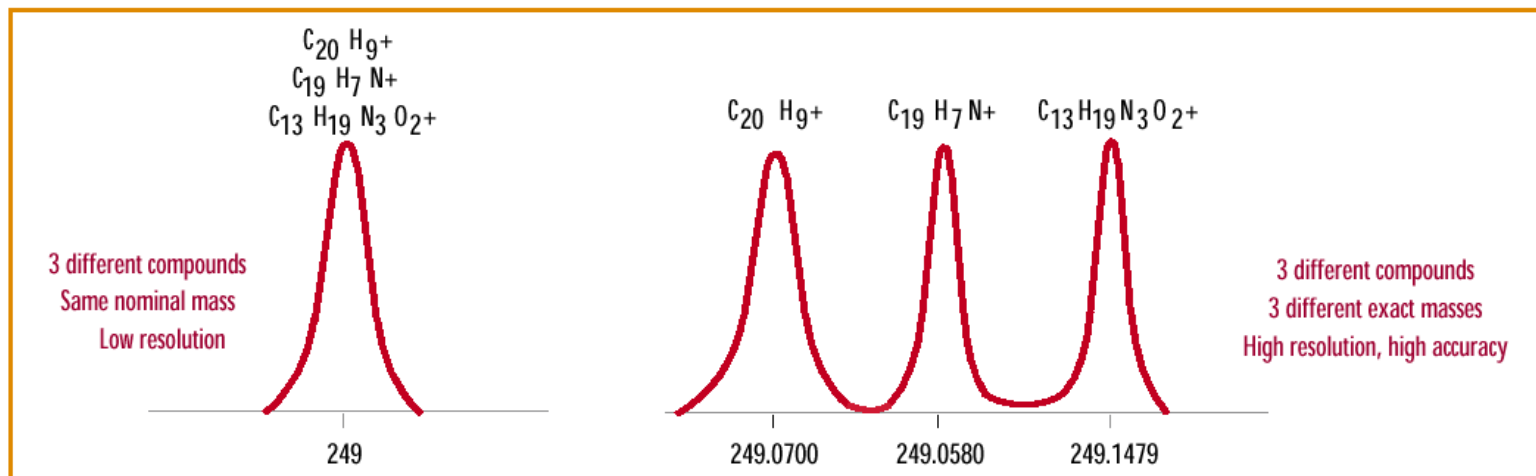
Figure 3.6. Resolution of the isotope cluster of a peptide. (A) Mass analysis of a triply charged peptide ion at a mass resolution of 6500. This resolution is sufficient to resolve the components of the isotope cluster and to measure the monoisotopic mass. (B) Mass analysis of a triply charged peptide ion at a mass resolution of 500. This resolution cannot resolve the isotope cluster of the ion, and the average mass of the peptide ion is determined.







High Mass Accuracy Measurements



Mass Accuracy

- Mass accuracy in *parts per million* (ppm)

$$\frac{(\text{measured mass} - \text{observed mass})}{\text{true mass}} \times 10^6$$

- Measured mass
 - 545.4200
- Known (calculated) mass
 - 545.3234
- Mass accuracy = $(545.4200 - 545.3234) / 545$
= 0.00011724 Da

$$= 0.00011724 \times 10^6 = 117 \text{ ppm}$$

A monoisotopic mass can be measured as accurately as the instrument allows, as long as the monoisotopic peak has been correctly identified. If the wrong peak is selected, the mass value will be false by one or two Daltons.

